## $\beta$ -lactams via photochemical reaction of monothioimides

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Summary: Photolysis of acyclic monothioimides gave  $\beta$ -lactams and thioamides via Type II reaction involving  $\gamma$ -hydrogen abstraction by thiocarbonyl group.

Photochemical reactions of thioketones and thioesters were well studied. However, those of thiocompounds corresponding to amides or imides are a little. Previously we reported some types of photochemical reactions of acyclic mono- or dithioimides, such as,  $\delta$ -hydrogen abstraction of N-thioacyl(thio)ureas, [2+2]photocycloaddition of N-thiobenzoylmethacrylamides,<sup>4</sup> and  $\beta$ -hydrogen abstraction of monothioimides. Now we wish to report a photochemical synthesis of  $\beta$ -lactams involving  $\gamma$ -hydrogen abstraction of acyclic monothioimides.

When N-isobutyroyl-N-phenylthiobenzamide (<u>1a</u>) was irradiated in benzene with a 1kw high pressure mercury lamp under argon, ir and nmr spectrum of the reaction mixture indicated the formation of  $\beta$ -lactam (2a) and thiobenzamide



1a

2a

38(29%) 5(70%)

(<u>4</u>). The  $\beta$ -lactam was too unstable to isolate as such. Benzoylation of the reaction mixture with benzoyl chloride and triethylamine gave S-benzoyl  $\beta$ -lactam (<u>3a</u>) in 29% and N-thiobenzoyl-N-phenylthiobenzamide (<u>5</u>) in 70%. The formation of <u>5</u> was reasonably explained in terms of benzoylation of <u>4</u> which was produced by Type II cleavage of <u>1a</u>. Photolysis of the monothioimide (1b)

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followed by benzoylation gave the  $\beta$ -lactam <u>3b</u> in 16% and <u>7</u> in 39%. The formation of <u>7</u> from <u>6</u> was confirmed by the fact that <u>7</u> was independently synthesized from <u>6</u> by benzoylation. In the case of <u>1c</u>, S-benzoyl  $\beta$ -lactams (<u>3c</u>, about 1:1 mixture of stereoisomers), S-phenylacetyl  $\beta$ -lactam (<u>3c'</u>), and <u>7</u> were obtained in the respective yields of 35%, 13%, and 17%.<sup>6</sup>



Photolysis of methoxy acetyl derivative (<u>1d</u>) gave  $\beta$ -lactam (<u>3d</u>) in high yield (76%) in the same manner. In this case, only cis isomer was obtained. Photolysis of <u>1e</u> followed by acetylation with acetyl chloride and triethyamine gave S-acetyl  $\beta$ -lactam (<u>3e</u>) as a mixture of two stereoisomers (cis, 63% : trans, 27%) in 90% yield.



For the mechanism, Type II reaction is postulated as shown below. The intermediacy of the zwitterion <u>9</u> is included by the following reason. Previously we reported the photochemical reaction of  $\alpha$ -oxoamides (<u>10</u>) which gave  $\beta$ -lactams (<u>12</u>) and oxazolidinones (<u>13</u>) via zwitterionic intermediate



11.<sup>7)</sup> Zwitterion 9 resembles 11. Zwitterion 9 may be more stable than 11 because of the sulfur atom at cation center and the absence of a hydroxy group at the anion center as compared with zwitterion 11. The formation of the ketenes (Type II cleavage products) was supported by the fact that S-phenyl-acetyl  $\beta$ -lactam (3c') was obtained in the photolysis of 1c. For the formation of 4 and 6, direct homolysis of C-N bond of the monothioimide (1) could not be excluded. In the present reaction, Type II cyclization occurred as a main process. On the other hand, Mazzocchi et al. reported the photochemistry of acyclic imides. The imides have been shown to undergo Type II eliminations across the imide moiety in addition to those on the C-alkyl chain and  $\alpha$ -cleavage reaction. The differences in the photochemical behavior between acyclic imides and acyclic monothioimides can not be explained at present.



In conclusion, acyclic monothioimides undergo photochemical  $\gamma$ -hydrogen abstraction to produce  $\beta$ -lactams (Type II cyclization products), ketenes, and thioamide (Type II cleavage products). The present reaction provides not only the first example of  $\gamma$ -hydrogen abstraction of thioimide derivatives but also a useful synthetic method of some  $\beta$ -lactams possessing sulfur atoms at the C-4 possition.

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